AMENDMENTS TO THE CLAIMS

Claims 1 - 10 (previously cancelled)

Claim 11 (currently amended) A process for preparing a defective metal oxide for a battery cathode with increased lithium capacity, said process comprising:

providing a sufficient amount of metal oxide;

heating said metal oxide under an atmosphere consisting essentially of O_2 and $H_2O_{(g)}$ gas wherein said heating step is maintained at a temperature of from about 300 to about 600 °C; and cooling said metal oxide, wherein said heating under said atmosphere introduces local ionic defects and increases the lithium capacity of said metal oxide.

Claims 12-16 (previously canceled)

Claim 17 (cancelled): The process as in claim 11, further comprising the step of: maintaining said heating at a temperature of from about 300 to about 600 °C.

Claim 18 (currently amended): The process as in claim <u>11</u> 17, wherein said heating is maintained from about 6 to about 72 hours.

Claim 19 (previously amended): The process as in claim 11, wherein said O_2 and $H_2O_{(g)}$ is applied to said metal oxide sample at a linear flow rate of about 50 ccm to about 350 ccm.

Claim 20 (previously amended): The process as in claim 11, wherein said heating is from about 2 to

about 20 °C/min up to about 460 °C.

Claim 21 (previously presented): The process as in claim 20, further comprising the step of:

maintaining said temperature of about 460 °C for 24 hours.

Claim 22 (previously presented): The process as in claim 11, wherein said cooling is from about 2

to about 20 °C/min until ambient air temperature is achieved.

Claim 23 (previously amended): The process as in claim 11, wherein said metal oxide is V₂O₅.

Claim 24 (previously amended): The process as in claim 11, wherein said metal oxide comprises a

surface area of about 1- 10 square meters.

Claim 25 (previously amended) A process for preparing a defective metal oxide for a battery

cathode with increased lithium capacity, said process comprising:

providing a sufficient amount of metal oxide;

heating said metal oxide at a temperature of from about 300 to about 600 °C for a time

period of from about 6 to about 72 hours under an atmosphere consisting essentially of O₂ and

 $H_2O_{(g)}$ gas, wherein said O_2 and $H_2O_{(g)}$ is applied to said metal oxide sample at a linear flow rate of

about 50 ccm to about 350 ccm; and

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cooling said metal oxide, wherein said heating under said atmosphere introduces local ionic defects and increases the lithium capacity of said metal oxide.

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